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[ion definition chemistry examples facts britannica](#) Dec 26 2022 ions are formed by the addition of electrons to or the removal of electrons from neutral atoms or molecules or other ions by combination of ions with other particles or by rupture of a covalent bond between two atoms in such a way that both of the electrons of the bond are left in association with one of the formerly bonded atoms

*what is an ion definition and examples thoughtco* Nov 25 2022 sep 4

2019 an ion is defined as an atom or molecule that has gained or lost one or more of its valence electrons giving it a net positive or negative electrical charge in other words there is an imbalance in the number of protons positively charged particles and electrons negatively charged particles in a chemical species history and meaning

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